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Mixed Ionic Covalent Compound Naming

Ionic/Covalent Compound Naming Solutions. For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately. 1) Na_2CO_3 sodium carbonate. 2) P_2O_5 ...

Mixed ionic and covalent naming worksheet and answers

...

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Write the names of the following compounds: 1) CH₄ methane
2) Mg(HCO₃)₂ magnesium bicarbonate (magnesium hydrogen carbonate)
3) CO carbon monoxide 4) P₄ phosphorus 5) SiS₂ silicon disulfide
6) PBr₃ phosphorus tribromide 7) Ag₃PO₄ silver phosphate
8) Si₃N₄ trisilicon tetranitride 9) CoCl₂ cobalt (II) chloride
10) Ga₂S₃ gallium sulfide

Mixed Ionic & Covalent Naming

Start studying Mixed Ionic/Covalent Compound Naming. Learn vocabulary, terms, and more with flashcards, games, and other study tools.

Mixed Ionic/Covalent Compound Naming Flashcards | Quizlet

For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately. 1) Na₂CO₃ sodium carbonate, 2) P₂O₅ diphosphorus pentoxide. 3)

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1) I₂ Iodine. 2) FeSO₄ iron (II) sulfate. 3) SiO₂ silicon dioxide. 4) GaCl₃ gallium chloride. 5) CoBr₂ cobalt (II) bromide. 6) B₂H₄ diboron tetrahydride. 7) CO carbon monoxide

Mixed Ionic/Covalent Compound Naming

Ionic/Covalent Compound Naming Solutions For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately. 1) Na₂CO₃ sodium carbonate 2) P₂O₅ diphosphorus pentoxide 3) NH₃ ammonia 4) FeSO₄ iron (II) sulfate 5) SiO₂ silicon dioxide 6) GaCl₃ gallium chloride 7) CoBr₂ cobalt (II) bromide 8) B₂H₄ diboron tetrahydride 9) CO carbon monoxide 10) P₄ phosphorus For each of the following questions, determine whether the compound is ionic ...

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Naming Mixed Ionic and Covalent - Answers. Name the following compounds. Remember, they may be either ionic or covalent compounds, so make sure you use the right naming method! 1) NaF sodium fluoride. 2) NF₃ nitrogen trifluoride. 3) Li₂O lithium oxide. 4) Al₂S₃ aluminum sulfide. 5) MgSO₄ magnesium sulfate. 6) SiH₄ silicon tetrahydride. 7) KNO₃

Naming Mixed Ionic and Covalent Compounds

Ionic & Covalent Compound Naming First, identify whether these compounds are ionic or covalent. Then, use the correct naming rules to write the correct names for each compound. Chemical

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Formula Type of Compound: Ionic or Covalent Compound Name
21) CdBr₂ 22) Cr₂O₇ 23) SBr₂ 24) (NH₄)₂CrO₄ 25) CuO 26) Pt

Ionic & Covalent Compound Naming Race

Steps to Naming Covalent Compounds. First, identify the elements present. Second, look at the subscript of each element to determine which prefix to use. (If an element does not have a prefix, assume that the subscript is "1."). Third, apply the above naming scheme.

Steps to Naming Ionic and Covalent Compounds - Owlcation ...

Ionic/Covalent Compound Naming Solutions 1) Na₂CO₃ sodium carbonate 2) P₂O₅ diphosphorus pentoxide 3) NH₃ ammonia 4) FeSO₄ iron (II) sulfate 5) SiO₂ silicon dioxide 6) GaCl₃ gallium chloride 7) CoBr₂ cobalt (II) bromide 8) B₂H₄ diboron

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tetrahydride 9) CO carbon monoxide 10) P 4 phosphorus 11) dinitrogen trioxide N₂O₃ 12) nitrogen N₂

Naming Ionic Compounds Practice Worksheet

Naming Mixed Ionic and Covalent - Answers Name the following compounds. Remember, they may be either ionic or covalent compounds, so make sure you use the right naming method! 1) NaF sodium fluoride 2) NF₃ nitrogen trifluoride 3) Li₂O lithium oxide 4) Al₂S₃ aluminum sulfide 5) MgSO₄ magnesium sulfate 6) SiH₄ silicon tetrahydride 7) KNO₃

Naming Mixed Ionic and Covalent Compounds

Diphosphorus pentoxide (covalent) K₂O. Potassium oxide (ionic) SF₃. Sulfur trifluoride (covalent) Fe₂O₃. Iron (III) oxide (ionic) CoCl₂. Cobalt (II) chloride (ionic)

Naming Chemical Formulas (mixed) Flashcards | Quizlet

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Ionic/Covalent Compound Naming Solutions For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately. 1) Na_2CO_3 sodium carbonate 2) P_2O_5 diphosphorus pentoxide 3) NH_3 ammonia 4) FeSO_4 iron (II) sulfate 5) SiO_2 silicon dioxide 6) GaCl_3 gallium chloride 7) CoBr_2 cobalt (II) bromide 8) B_2H_4 diboron tetrahydride

Mixed Ionic/Covalent Compound Naming

N_2O_3 is named dinitrogen trioxide because it contains two nonmetals, not because the bonds have high covalent character. FeO or Fe_2O_3 is named either iron (II) oxide or iron (III) oxide because it...

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Naming Covalent Molecular Compounds Worksheets -

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Kiddy Math

Chapter 6 Ionic and Covalent Compound Naming (Practice Quiz) (with oxidation numbers and correct subscript latex codes) Take and pass with 70% for 5 point bonus on your test.

Quia - Chapter 6 Ionic and Covalent Compound Naming ...

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